

Estimation of Thiocyanate Content from Selected Cruciferous Vegetables

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As per the studies conducted by the Indian Thyroid Society, it is suggested that one out of ten people in India suffer from hypothyroidism and these figures are likely to increase in the near future. According to the survey, women in the post-menopausal age group are more affected in comparison to men. Globally, the prevalence of hypothyroidism is 4-5%. Taking the same into consideration, a comparative study was undertaken to estimate the thiocyanate content in cruciferous vegetables such as *Raphanus sativus* (radish), *Brassica rapa* (turnip), *Brassica oleracea* var. botrytis (cauliflower) *Brassica oleracea* (broccoli), *Brassica clearcea* var. botrytis (cabbage). Increased levels of thiocyanate often cause imbalance of the thyroid hormones by inhibiting iodine uptake thereby leading to hypothyroidism. Thiocyanate is therefore also employed in the treatment of hyperthyroidism. This study used 10 gram of sample extract in triplicates that were obtained from different markets, treated with ferric chloride leading to the formation of a colored complex and the amounts of thiocyanate was evaluated colorimetrically. It was observed that the thiocyanate content in cabbage was the highest followed by radish, turnip, broccoli and cauliflower. High thiocyanate content coupled with low iodine content go a long way in leading to goiter which is a clinical manifestation of any abnormality associated with thyroid gland. This leads us to very important conclusion that while ascertaining causes of any abnormalities of the thyroid gland that a person's diet needs to be taken into consideration.

Keywords: Goiter; Goitrogen; Hypothyroidism; Post-menopausal; Thiocyanate.

Thyroid diseases are one of the most widespread diseases in the world. The same holds true for the Indian scenario. Studies suggest an increasing incidence of thyroid disorders. As per a meta-analysis conducted by Anne and Rahiman (2022) it was concluded that 1 in 1031 neonates in India, suffered from congenital hypothyroidism which is significantly higher than developed countries like Japan, Germany and United Kingdom where the prevalence is 1 in 2500 to

3500, 1 in 3330 and 1 in 1887 respectively.¹ As per the guidelines set by the American Thyroid Association (2023), the RDA for iodine is 150µg/day for healthy adults.²

Thyroid hormones are of vital importance in the normal metabolism and virtually affect every cell of the body. The thyroid gland is found in the neck and consist of two lobes connected by the isthmus.² It produces two related hormones 3,5,3',5'-tetraiodothyronine (T4) and

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3,5,3'-triiodothyronine (T3). Tyrosine an amino acid that is synthesized in the body in sufficient amounts, is the major substrate that combines with iodine to form these hormones. Unlike tyrosine that is produced in the body³, iodine is primarily supplied through diet especially in meat products that contain up to 400 µg/kg.⁴ Iodine that is absorbed in the blood stream is taken up by several glands such as the salivary glands and thyroid glands in the form of I⁻ using the Na⁺/I⁻ symporter which is further oxidized to iodine.⁵

Thyroid hormones are known to have a pronounced effect on the metabolism of various biomolecules. It was reported that increased levels of thyroid hormones increase the rate of glucose metabolism by increasing the transcription of insulin; a hormone necessary for absorption of glucose by the body cells. Increased levels of insulin lead to increase in its secondary effects on the overall body.⁶ In case of increased thyroid hormone levels, the rate of glycogenolysis and gluconeogenesis also increase. It can thus be concluded that increased thyroid hormone levels lead to a hyperglycemic state. It was pointed out that since the metabolism of glucose and thyroid functions are interlinked it is implausible to rule out the connection of iodine deficiency and diabetes mellitus.⁷

It was reported that T3 and T4 stimulate lipolysis owing to an increased mobilization of triglycerides stored in adipose tissue. Increased levels of Thyroid Stimulating Hormone (TSH) are positively correlated to both, Low Density Lipoproteins (LDL) and total cholesterol levels in serum.⁸

A study conducted by Ibad *et al.*, (2023) suggested that increased levels of free thyroxine in the blood stream can be positively correlated to sarcopenia which is indicative of their catabolic effect at higher concentration.⁹

A study carried out by Yoo *et al.*, (2021) inferred that local regulation of thyroid hormones in the Central Nervous System may physiologically regulate appetite. Increased thyroid hormone levels may increase the metabolic rate thereby increase the appetite of an individual whereas decrease in the thyroid levels may slow down the basal metabolic rate thereby decreasing the appetite of an individual.¹⁰

The increase in the levels of thyroid

hormones both T3 and T4 is known as hyperthyroidism conversely a decrease in their levels is known as hypothyroidism.

The diseases that are chronically related to the thyroid gland are hyperthyroidism, Hashimoto's thyroiditis,¹¹ Grave's disease,¹² cretinism¹³ to name a few. Goitre is the clinical manifestation of any abnormality associated with the thyroid gland.

A myriad of chemical compounds in the environment and some drugs are known to hinder the functions of the thyroid gland thus posing the danger of diseases of thyroid gland.¹⁴ Further, substances found in the environment such as water, food sources that lead to goitre are called environmental goitrogens whose exposure or consumption should be limited as their intake over an acceptable limit can be deleterious to health.¹⁵ Some of the examples of goitrogens are thiocyanates, perchlorates, thiourea sulfamethoxines.

Perchlorates are substances used in the development of various explosives pose severe threat to the thyroid gland as they act as a competitive inhibitor of iodide thereby hindering its uptake.¹⁶ Recently, perchlorates have come under great inspection as studies report that there is a greater environmental distribution especially in potable water than what was previously known.¹⁷ Based on the study conducted by Legakis *et al.*, (2022) exposure to perchlorates through smoking, radiation may increase the risk of thyroid cancer that accounts for about 90% of all endocrine cancers¹⁸

Thiourylenes inhibit the iodination of monoiodotyrosine and thereby block the coupling reaction. A study concluded that 6-n-propylthiouracil which is a derivative of thiourea is a safe and potent drug for treatment of hyperthyroidism. Clinically used drugs such as propylthiouracil and methimazole cause hindrance in the iodination of tyrosine. They act as competitive inhibitors of tyrosine and themselves get iodinated. Further propylthiouracil also inhibits 5['] DI thereby reducing the conversion of T4 to T3 in many extra-thyroid tissues.¹⁹ Drugs such as methimazole and carbimazole cause inhibition of the enzyme thyroperoxidase.²⁰

Thiocyanates are anions that have antithyroid activity. They inhibit the uptake of iodine by acting as competitive inhibitors of

the sodium/iodine symporter.²¹ It thus worsens iodine deficiency by inhibition of thyroidal uptake of iodine. Thiocyanate is found in large concentration in the cruciferous vegetables.²² It is found in its progoitrogen form as glucosinolates which on cleavage by myrosinase enzyme leads to the production of thiocyanate.²³ A recent study reported that thiocyanate exposure along with other endocrine disruptors even in low concentration could alter the thyroid hormone homeostasis in adults in China.²⁴

The primary objective of this research is to compare the thiocyanate content in different cruciferous vegetables based on purchases from various markets of the city. The vegetables included were radish, turnip, cauliflower and cabbage. Various established, sensitive and specific methods were tested for the estimation of thiocyanate content but out of all, the modified Johnston and Jones method proved convenient. This method employed the formation of a red coloured complex of iron-thiocyanate whose extinction was measured at 530nm.

Various other techniques are available one of which, involves formation of a pink coloured pyridine-benzidine complex in the presence of thiocyanate and the absorbency is measured at 525nm. This method was employed to estimate the concentration of thiocyanate in urine.²⁵

A colorimetric estimation was carried out by Duch *et al.*, (2019) which measured a red

Table 1. Instruments used

Instrument	Name of the Company
Colorimeter	Equiptronics

Table 3. Thiocyanate content in vegetables per 100gm wet weight obtained from six different markets

Vegetables	Thiocyanate content obtained from different markets (mg/100gm)					
	Market 1	Market 2	Market 3	Market 4	Market 5	Market 6
Turnip	1.72	1.82	1.72	1.72	1.82	1.82
Cabbage (Outer leaves)	2.51	2.32	2.42	2.12	2.68	2.64
Cabbage (Inner leaves)	2.22	2.28	2.16	1.85	2.48	2.42
Radish	2.42	2.35	2.42	1.92	2.51	2.37
Cauliflower	0.86	1.13	0.71	1.23	1.73	1.29
Broccoli	1.42	2.07	1.92	1.95	1.66	2.42

The values are mean \pm S.D of 3 samples obtained from a particular market in Mumbai.

coloured complex concentration that was a result of $\text{Fe}(\text{SCN})_6^{3-}$ complex formation when the extract was treated with trichloroacetic acid and ferric ions.²⁶

A similar study conducted for finding the concentration of thiocyanate colorimetrically made use ferric nitrate which yield a complex whose absorbance was measured at 470nm.²⁷

MATERIAL AND METHOD

Materials

Instruments: The instruments that had been used are as follows:

Reagents: Potassium Thiocyanate [KCNS]: Working standard of KCNS (30 $\mu\text{g}/\text{mL}$) was prepared and serially diluted with distilled water to obtain concentrations of 5 $\mu\text{g}/\text{mL}$, 10 $\mu\text{g}/\text{mL}$, 15 $\mu\text{g}/\text{mL}$, 20 $\mu\text{g}/\text{mL}$, 25 $\mu\text{g}/\text{mL}$ and 30 $\mu\text{g}/\text{mL}$. Standard Iron chloride [FeCl_3] of concentration 0.4M was prepared in distilled water. 5% mercuric chloride was prepared in 1N HCl. All chemicals used were of AR/GR grade.

Table 2. Dilution table for blank, standard and unknown tubes

Tube no.	Working Standard (0.03mg/mL)	Distilled Water (mL)	FeCl_3 (mL)
Blank	-	3.0	1.5
1	0.5	2.5	1.5
2	1.0	2.0	1.5
3	1.5	1.5	1.5
4	2.0	1.0	1.5
5	2.5	0.5	1.5
6	3.0	-	1.5
Unknown	3.0	-	1.5

Methods

Sample preparation

Vegetable samples were obtained from 6 different local vegetable markets of Mumbai. 3 samples of each vegetable were acquired from every market. Therefore, a sample size of 18 for every vegetable was procured for analysis.

Each sample was taken in triplicates (10g each), crushed, extracted with 10 mL distilled water and refluxed for 45 minutes in a conical flask. The refluxed samples were allowed to cool until room temperature was obtained and were then filtered using muslin cloth and filter papers. The residues were given repeated washings with distilled water

Table 4. Average thiocyanate content in vegetables per 100gm wet weight

No.	Sample	Thiocyanate (-SCN)(mg/100gm)
1.	Turnip <i>Brassica rapa</i>	1.77±0.009
2.	Radish <i>Raphanus sativus</i>	2.33±0.03
3.	Cabbage <i>Brassica cleracea var. capitata</i>	
	(a) Inner leaves	2.23±0.03
	(b) Outer leaves	2.44±0.03
4.	Cauliflower <i>Brassica okracea var. botrytis</i>	1.13±0.05
5.	Broccoli <i>Brassica oleracea</i>	1.90±0.05

The values are mean ± S.D of 6 samples obtained from different markets in Mumbai.

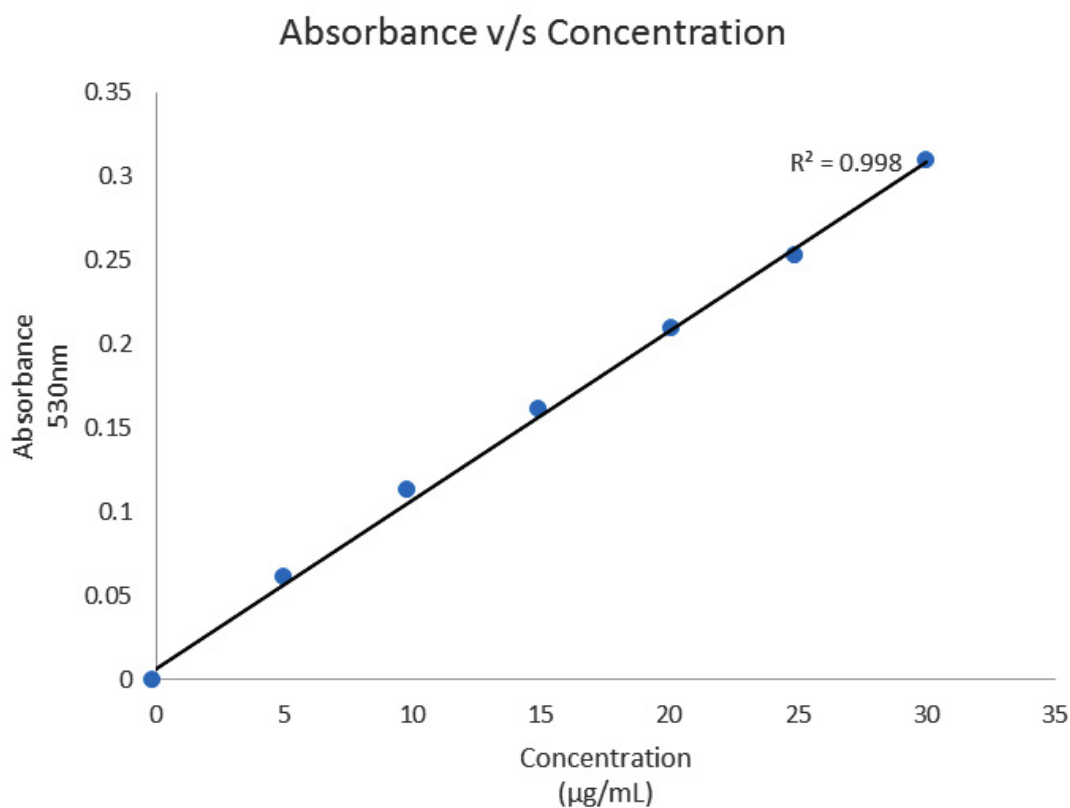


Fig. 1. Standard curve of thiocyanate concentration, Absorbance at 530 nm against concentration (µg/mL)

and the volumes of the filtrates were made up to 50mL.

In order to negate the green colour imparted to the extract of certain vegetables, 6mL of the elute was taken and distributed in 2 different test tubes. 2-3 drops of 5% mercuric chloride were added to one of the tubes to it to block thiocyanate.

Ferric chloride was then added to the tubes and the extinction was measured.

Estimation of thiocyanate

Thiocyanate estimation was done using modified Johnston and Jones method. This method involves the formation of a red colour complex which is a result of the reaction between SCN⁻ ions

Comparison of thiocyanate content from turnips obtained from different markets

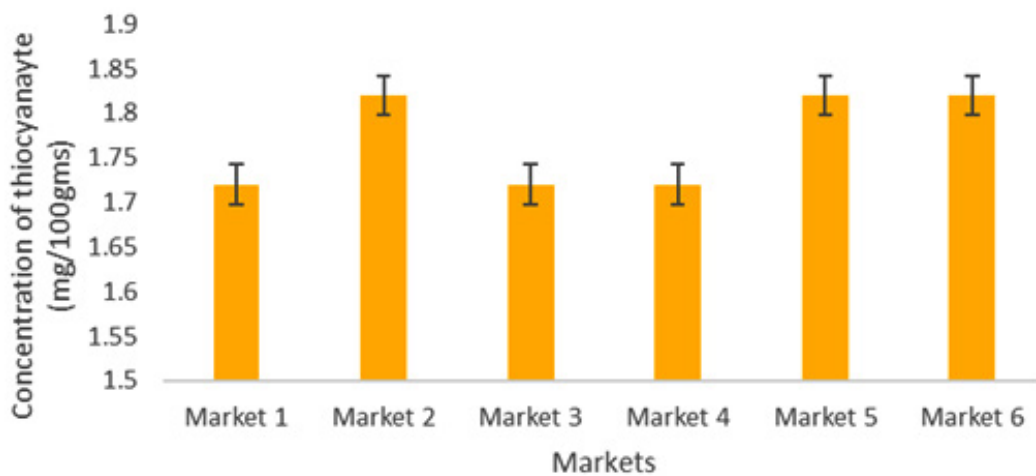


Fig. 2. Comparison of thiocyanate content in turnips obtained from different markets

Comparison of thiocyanate content from cabbage (outer leaves) obtained from different markets

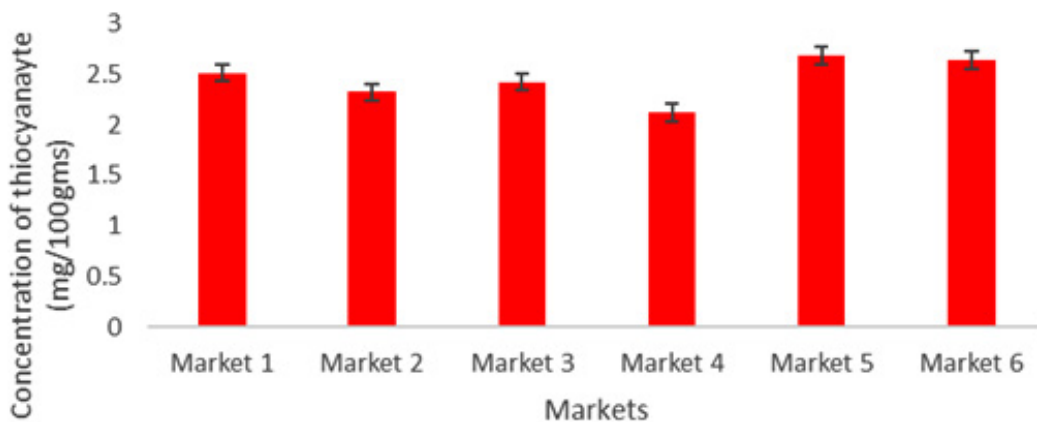
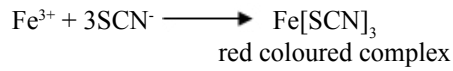


Fig. 3. Comparison of thiocyanate content in cabbage (outer leaves) obtained from different markets

from the vegetable sample and Fe from FeCl₃ in the +3 oxidation state.



The extinction values of blank, standard and test samples were measured at 530nm. The concentrations of the unknown samples were computed from the standard curve. The series of standard tubes were prepared in the following manner:

Comparison of thiocyanate content from cabbage(inner leaves) obtained from different markets

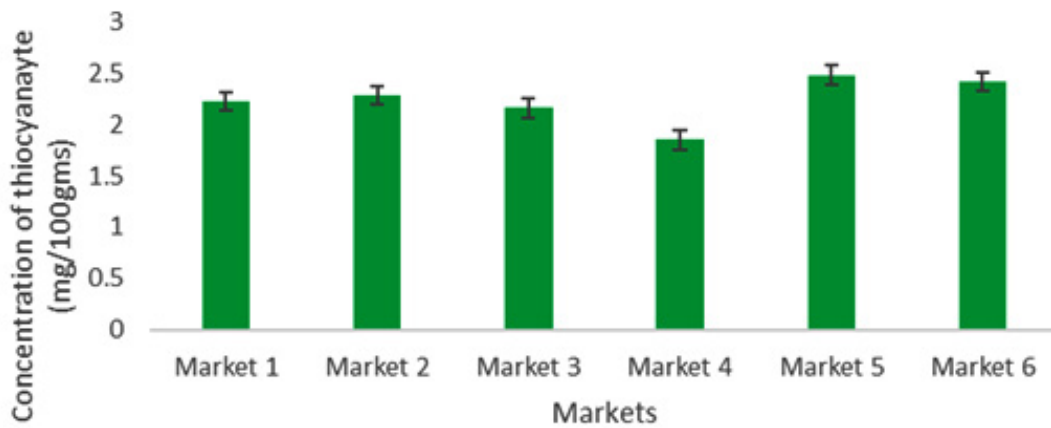


Fig. 4. Comparison of thiocyanate content in cabbage (inner leaves) obtained from different markets

Comparison of thiocyanate content from raddish obtained from different markets

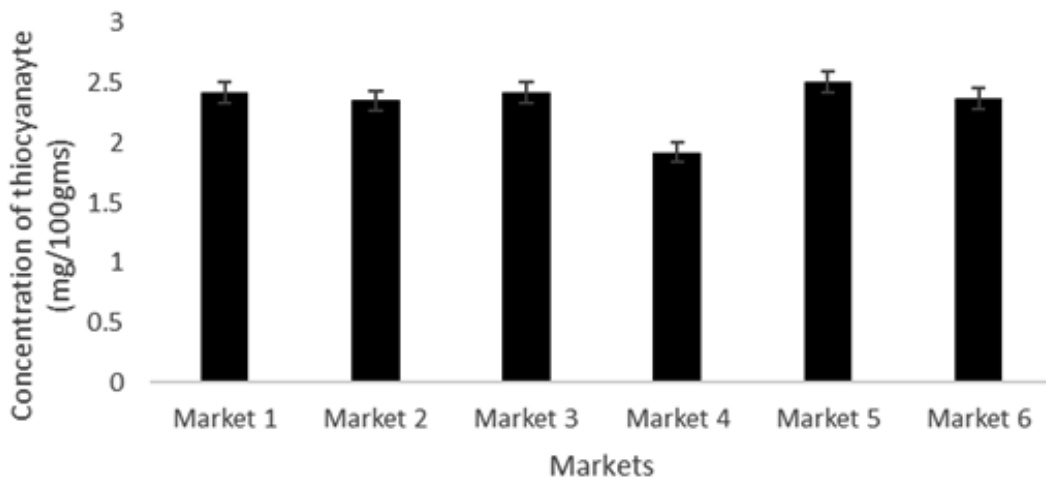


Fig. 5. Comparison of thiocyanate content in raddish obtained from different markets

RESULTS

The amount of thiocyanate from various vegetables was determined using a standard curve in which the value obtained in $\mu\text{g}/10\text{g}$ has been scaled up to report the value in $\text{mg}/100\text{gm}$. A wide range of thiocyanate content was observed

across different cruciferous vegetables. The lowest mean level of thiocyanate yield was observed in cauliflower ($2.44 \pm 0.03 \text{ mg}/100\text{gm}$) as compared to the other vegetables chosen for the experiment. The difference in the thiocyanate concentration of the outer and inner leaves of cabbage may be attributed to the fact that the outer leaves are more mature

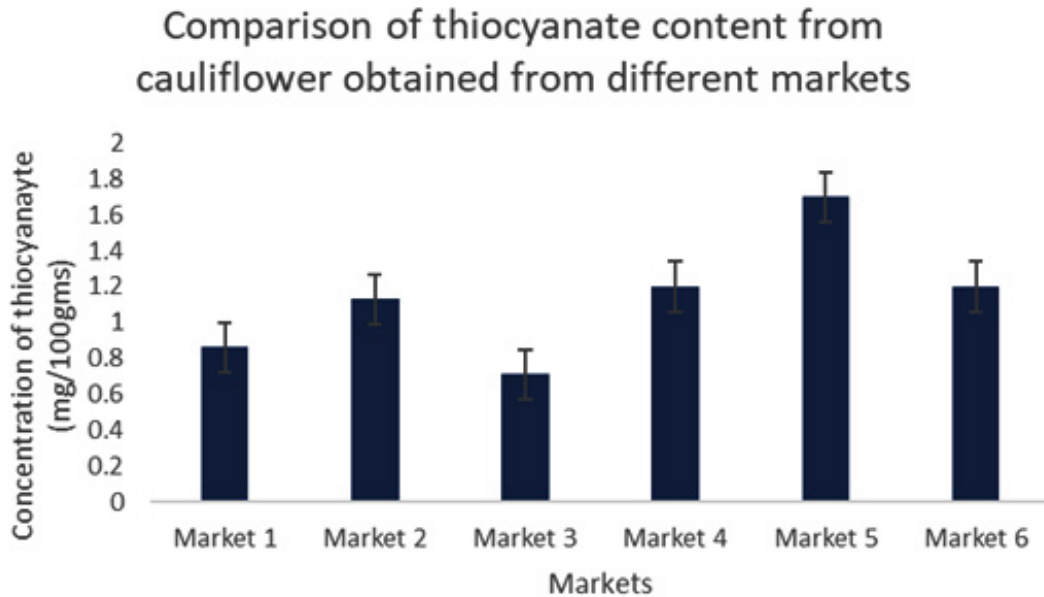


Fig. 6. Comparison of thiocyanate content in cauliflower obtained from different markets

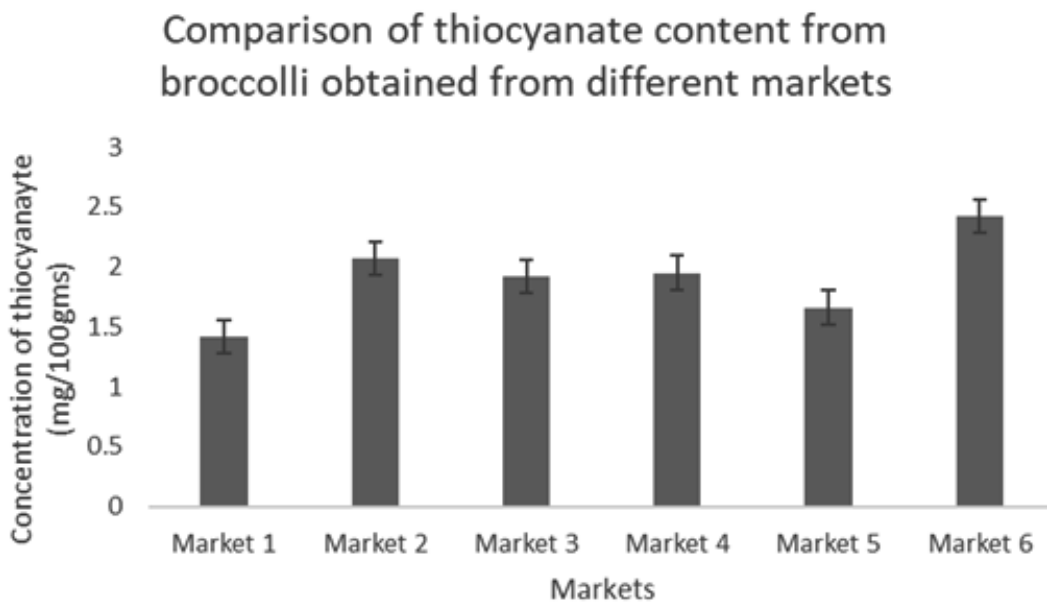


Fig. 7. Comparison of thiocyanate content in broccoli obtained from different markets

as compared to the inner leaves. The average thiocyanate content in the selected vegetables is as depicted in table 4. In order to negate the bias of sampling, we compared the total thiocyanate content of vegetables that were purchased from six different local markets.

A two-way ANOVA was conducted to check for the difference in the thiocyanate concentration for different vegetables obtained from different markets in Mumbai. There is a statistically significant difference seen in the thiocyanate concentration of different vegetables ($F = 29.008$, $p\text{-value} = 1.204 \times 10^{-9}$). Thus, the null hypothesis is rejected. The concentration of thiocyanate seen in different vegetables is statistically significant.

There is a statistically significant difference seen in the thiocyanate concentration of samples of any particular vegetable obtained

from 6 different markets ($F = 2.7315$, $p\text{-value} = 0.042$). Thus, the null hypothesis was rejected. The thiocyanate concentration of a particular vegetable obtained from 6 different markets in Mumbai is statistically significant.

DISCUSSION

The current study aimed to evaluate the thiocyanate concentration of different cruciferous vegetables. The method employed was a modified Johnston and Jones method. A similar colorimetric estimation conducted in Poland by Przybylska *et al.*, (2022) reported that cabbage has a thiocyanate concentration of 1.194 ± 0.001 mg/100gm whereas broccoli has a thiocyanate concentration of 1.034 ± 0.001 mg/100gm. These results match the findings of our study as they suggest that cabbage has a higher concentration of thiocyanate as

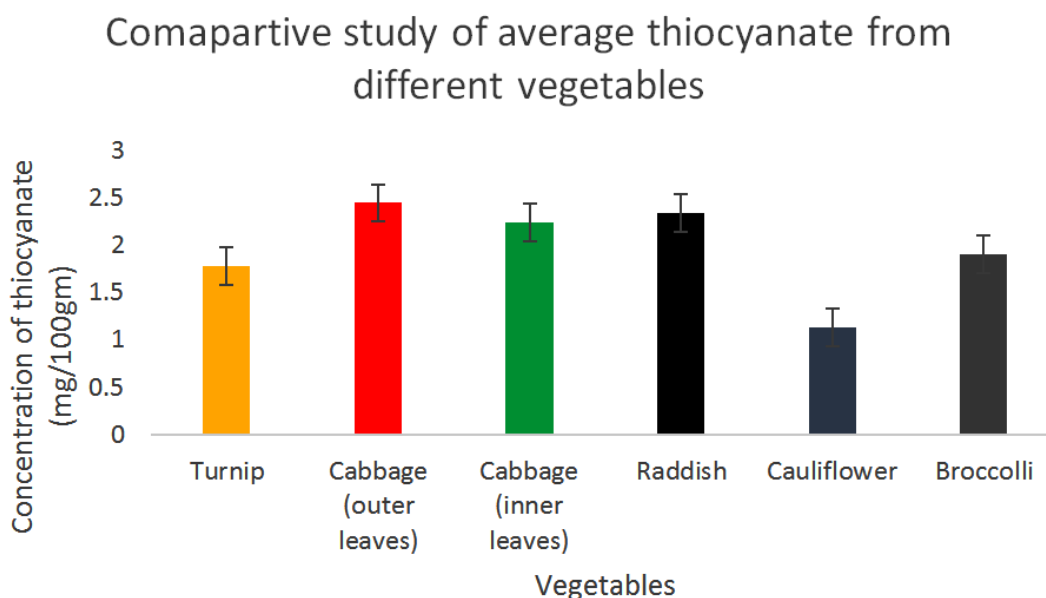


Fig. 8. Comparative study of average thiocyanate content from different vegetables

Table 5. Two-way ANOVA test for the thiocyanate concentration of different vegetables obtained from 6 markets in Mumbai

Source	DF	Sum of Square (SS)	Mean Square (MS)	F Statistic (df_1, df_2)	P-value
Factor A – Vegetables	5	7.0431	1.4086	29.008 (5,25)	1.204e-9
Factor B - Markets	5	0.6632	0.1326	2.7315 (5,25)	0.04212

compared to broccoli.²⁸ The results of our study were also in line with a similar study conducted by Sun and Chen in the year 2019 on members of the cruciferous family which suggests that broccoli has a higher thiocyanate concentration than cauliflower.²⁹ The study conducted on Allium vegetables by Czech *et al.*, (2022) such as garlic, yellow onion, red onion, and leek reported a lower thiocyanate concentration as compared to the vegetables of the cruciferous family included in our study which suggests that thiocyanate content is found more in members of the cruciferous family.²⁷ Differences in thiocyanate yield amongst individual cruciferous vegetables may be chalked up to environmental factors and hereditary factors, which determine the content of glucosinolates and/or the ratio of thiocyanate-producing glucosinolates to the other glucosinolates in cruciferous vegetables as reported by Liu *et al.*, (2020).³⁰ Differences in thiocyanate concentration found in different vegetables in our study can be attributed to the difference in glucosinolates as a study conducted by Abbaoui *et al.*, (2018) reported that different concentrations in glucosinolates yield different amounts of isothiocyanates, thiocyanates and other metabolites.³¹

CONCLUSION

In this study, the vegetables examined include members of the cruciferous family that are very integral to Indian cuisine. Thus, the data obtained from this study can be used to assess the exposure of the population to thiocyanate especially through their diet. It was reported that repeated intake of small quantities of these vegetables over a prolonged period of time have a negatory effect on the thyroid gland, especially coupled with a low iodine diet.²⁵ Repeated consumption of thiocyanate through diet can lead to an increase in the urinary thiocyanate levels which can lead to a decrease in the triiodothyronine levels.³² Also, thiocyanate can act as a competitive inhibitor of sodium/iodine symporter²⁰, in high concentrations can lead to decrease in thyroid hormones synthesis and could also inhibit the uptake of iodine in infants, resulting in a decrease iodine available to them.³³ A reduction in the regular consumption of thiocyanate can thus go a long way in treatment of hypothyroidism. Due to the extreme fluctuations in the concentration of

iodine found in different sources of food and water, it was suggested that in order to combat iodine deficiency, making salt iodination mandatory is the best way moving forward.³⁴ Although several national and international endeavours have been undertaken to increase the intake of iodine either volitionally or through mandated iodination of salt and have achieved success in many countries, the problems of iodine deficiency still remain in countries like Russia, Australia, Africa and few countries of Asia.³⁵ Therefore, from this study we effectively conclude that when ascertaining causes of any thyroid gland abnormalities, an individual's diet needs to be taken into consideration.

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Conflict of interest

The authors declare that there are no conflicts of interest in the course of conducting the research. All the authors had final decision regarding the manuscript and decision to submit the findings for publication.

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